

STRANGE POWER OF RADIIUM

MOST CONCENTRATED FORM OF ENERGY KNOWN.

A Lecture by Sir William Ramsay Telling What is Known of Radium and its Emanations—Three Kinds of Rays—Fourative Properties of Radium.

LONDON, Feb. 16.—Speaking at the Auditors' Club last Monday night on the subject of radio-activity Sir William Ramsay summed up nearly everything known about radium and its emanations.

Starting with a definition of the terms concentration and energy, he traced briefly the discovery of radium by Mme. Curie and the methods of its extraction from uranium ores. Then after a description of the cathode rays and the discovery of M. Becquerel, he pointed out how all material human progress really consists in the first place in the concentration of energy, and in the second place in raising what was termed the economical coefficient.

Becquerel found that there were three different kinds of rays given out from radium. First there are rays which can be easily kept in a glass tube, which are known as Alpha rays. Then come a second kind, called Beta rays, which can pass through glass, and which turn out to be the same as the cathode rays of Lenard. The third kind, or Gamma rays, are very penetrating and appear to have some analogy to light, except that unlike light they can pass through non-transparent substances.

These rays can pass through about six inches of lead, and possess the power of discharging an electroscope. Lead, in fact, is moderately transparent to the Gamma rays. It was not quite clear whether the word ray was quite applicable to these rays or emanations, as there was a doubt as to whether they were really given off or were simply undulations or waves in the ether.

He should say that the Alpha and Beta rays were sometimes called particles, and not rays from the radium, whereas the Gamma rays corresponded more to waves in the medium which surrounds us and which is termed ether. As regards the Alpha rays, Prof. Rutherford was the first to point out that these were due to a gas, but a similar observation was made by a German, Dr. Schmidt, in connection with thorium, another element discovered by the speaker to be capable of discharging the electroscope when placed near it.

The fan, radiated, that radium was continually giving out gas. That was why the Alpha rays could get out of a vessel containing radium. The Beta rays were also particles—smaller than the Alpha rays, and travelling at an enormous velocity. The Gamma rays were not particles at all, but waves or undulations in the ether. These latter rays were still being investigated.

It was possible in some degree by means of calculation to measure the masses of the particles of the Alpha and Beta rays. Suppose a cannon to be pointed horizontally and loaded with a certain known charge of powder and a shot, if the cannon was fired the shot would not go in a straight line, but would describe a parabola, and the distance to which the shot would travel would depend upon its mass, upon the explosive force of the powder, and upon the attraction of the earth in dealing with the Alpha particles, and find out how much it was deflected from a straight line by the attraction of the electric field, and ascertain the rate at which it was propelled, they could calculate how heavy it was, and what its mass was.

Prof. Rutherford and Sir J. Thomson made such an experiment, and found that while the Alpha particles—that is, the gas which escapes from radium—have a very considerable mass, something about twice that of an atom of hydrogen, the mass of the Beta particles was exceedingly small, and was only equal to about one-thousandth of the mass of a hydrogen atom.

The Beta rays were being investigated and it had been found that they play an important part in the composition of matter. If matter was not made up of Beta particles revolving round each other like the planets of the solar system, something of the order of a Sir J. Thomson's little model of a planet revolving round each other. Returning to the Alpha particles Sir William proceeded to say that when radium emitted these particles it lost energy. It was always losing energy and it could not be stopped from doing that. It was true it went on very slowly and it was an interesting question how long it would take to emit all the Alpha particles which it contained. The first question was: What were these Alpha particles? The second was: Into what did radium change after emitting these particles? The third question was: How long did it take to undergo these changes?

The first change appeared to be into this gas—the so-called Alpha rays. Quite suddenly the energy of the radium was transferred to the gas which it gave off. This raised the question of the transmission of an element.

It was long believed to be possible to change one element into another. He was not going to throw cold water on the alchemists, because it was possible by the adding of small quantities of certain substances to change altogether the appearance of certain metals; for instance, by the addition of a small quantity of iron to a solution of copper into a white, brittle substance altogether unlike the malleable copper which was there before, and which you could change a large quantity of sugar into alcohol. These were changes of an alchemical nature, and the alchemists were continually doing what the alchemists of old tried to do when they attempted to change various elements into gold or silver. Radium went on changing and you could not stop it.

Prof. Rutherford gave the name of emanations to the rays given off by radium because he did not know what they were as gases, but it really was ordinary gas, or rather quite an extraordinary gas. It came off radium as a regular rate, and that brought him to the question: How long would radium last? His answer was, forever.

The amount of gas was always proportional to the amount of radium present. Suppose he took a slice of bread and cut it in two, taking a minute to do it, and then he cut the half in two again, and so went on continually cutting each piece in half, how long would it take him to cut that bread entirely up? He could never do it. He would be always halving to infinity, and the operation would take him an eternity to perform. That was the case with radium. The amount of gas produced was always proportionate to the amount of radium there and it was always being produced.

When would radium be half gone? They had just measured it in his laboratory, and it would take 1,750 years, so that if any one invested in radium he would retain at least half his capital at the end of 1,750 years.

The Austrian Government some time ago entrusted him with about half a gramme, or one fifty-fifth of an ounce, of radium for private use. It did not belong to him, but its value was about 25,000. Less than a year ago Dr. Gray and himself performed the experiment of isolating the Alpha emanation of radium,

and they enclosed it in a fine glass tube much finer than the finest thermometer tube that ever was made. They compressed it and liquefied it.

When liquefied it came with a purplish light, though it was quite transparent, like water. It shone with a light of its own. When reduced to a temperature of about ninety centigrade it solidified, and then it glowed with an extremely brilliant light like a miniature aurora. The quantity they had was extremely small, but they ascertained its boiling point, its melting point and its specific gravity, and all this was done with a smaller quantity than the point of the finest needle. Of course they used a microscope.

This substance was the most concentrated form of energy known. It was a substance which went on changing into other things, to which various names were given. These substances were named Radium A, Radium B, Radium C, and so on up to Radium F. Some of them had a short existence, lasting only half an hour or three-quarters of an hour, and he had never seen them. He had seen Radium D—a substance which would be gone in about forty years. That gave one time to look at it. It was a substance rather dull looking, like lead, and that was nearly all he could tell them about it. There were other substances, probably like the polonium discovered by Mme. Curie.

During these emanations radium gave out a great deal of energy, generally manifested as light, but as a matter of fact radium kept itself hot. There was a great deal of heat generated, and he calculated, and it was found that it gave off about 3,500,000 times as much heat as would be given off by the oxygen-blowing pipes, which gave out about 2,000 degrees Centigrade. There they had, surely, the greatest concentration of energy ever known.

What did this energy do? It sent out the Alpha rays at a velocity of about 200 million miles an hour, and it naturally carried a great deal of energy. The Beta rays, although only about one-thousandth the size, also carried great energy, and they were enormous in quantity. He had measured this experiment five times. The experiments were still going on.

There were plenty of things they could do with this energy. They could decompose water and metallic substances, and in these decompositions they found elements produced which they did not imagine to exist in the substances so treated. For instance, in decomposing ordinary copper sulphate they were surprised to discover lithium in what remained, and no trace of sulphur. He had repeated this experiment five times. The experiments were still going on.

The phenomena which he had described were hardly capable of explanation. He had nothing to do with it. He could only describe it as an act of God. An act of God had been described by an eminent lawyer, who was a Scotian, as "I weighed 45 kilos, was nearly 47 years of age, and accustomed to eating daily an amount of food approximately equal to the so-called dietary standards. Realizing that the habits of a lifetime should not be too suddenly changed, a gradual reduction was made in the amount of protein food taken each day, resulting in a month or so in the complete abolition of breakfast, except for a small out-of-door lunch was taken at 1:30 P. M., followed by a heavier dinner at 7:30. The total intake was diminished as well as the protein constituents."

"There was no change to a vegetable diet, but there was and still is a distinct tendency toward the exclusion of meat in some measure, the appetite not calling for this form of food in the same degree as formerly. At first the lessening of the daily amount brought a slight discomfort, but it soon passed away and an improved physical condition was perceptible. A rheumatic trouble in the knee joint which had persisted for a year and a half entirely disappeared; sick headache, bilious attacks and other minor troubles also ceased their regular visits; a keener appetite and more acute sense of taste was developed and a greater liking for simple foods. In seven months I weighed fifty-eight kilos instead of sixty-five."

"During the ensuing summer the same simple diet was persisted in—a small cup of coffee for breakfast, a fairly substantial dinner at midday and a light supper at night. Two months were spent in Maine at an inland fishing resort, and by rowing a boat frequently six to ten miles in a forenoon, sometimes against head winds and without breakfast, and with less resultant fatigue and muscular soreness than in the old days. I discovered the mistake that it is impossible or unwise to do physical work on an empty stomach."

"Returning to New Haven, I found that my body weight was practically the same as before the summer vacation. During the first months of the experiment eight kilos were lost under the gradual change of diet; then the weight apparently remained stationary, from which it might fairly be assumed that the body had finally adjusted itself to the new conditions."

From his experiments with the professional men Prof. Chittenden decides that certain general conclusions seem to be justified by the results obtained. "A healthy man," he says, "whose occupation is such as not to involve excessive muscular work, but whose activity is mainly mental, rather than physical, though by no means excluding the latter, can live on a much smaller amount of protein food than is usually considered essential for life. Loss of mental or physical strength and vigor."

This means that the ordinary professional man who leads an active and even vigorous life, with the burden of care and responsibility, need not clog his system and inhibit his power for work by the ingestion of any such quantities of protein food as the ordinary dietetic standard call for."

After giving this blow to the popular conception that the hard working professional man in the amount of food required to sustain his life, Chittenden goes on to prove his point by giving the resultant figures of the experiments carried on with himself and four of his colleagues as subjects. "We are certainly surprised," he says, "in the assertion that the professional man can safely practise a physiological economy in the use of protein food equal to a saving of one-third to two-thirds the amount called for by existing dietary standards, and this without increasing the amount of non-nitrogenous food consumed. The amount of protein food can likewise be diminished in amount without detriment to health or strength when there is no call for great physical exertion."

The experiment with the second group included twenty men—of whom fourteen were privates, volunteers for the experiment, and six were regular army cooks, cook's helper, etc., representing a great variety of types as possible and different nationalities and temperaments. The experiment was conducted at New Haven under military discipline and constant surveillance. A gradual sifting reduced the number to thirteen.

Further, in describing the experiments which he conducted with himself as a member of the first group says: "I weighed 45 kilos, was nearly 47 years of age, and accustomed to eating daily an amount of food approximately equal to the so-called dietary standards. Realizing that the habits of a lifetime should not be too suddenly changed, a gradual reduction was made in the amount of protein food taken each day, resulting in a month or so in the complete abolition of breakfast, except for a small out-of-door lunch was taken at 1:30 P. M., followed by a heavier dinner at 7:30. The total intake was diminished as well as the protein constituents."

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EATING IS ONE-HALF HABIT

PROF. CHITTENDEN'S TESTS AND HIS CONCLUSIONS.

Experiments with College Professors, Soldiers and Athletes Showed That They Turned on Much Less Food Than They Were Accustomed to Eat.

In connection with the discussion of the price of food and as substitutes for meat some observations made by Prof. Russell H. Chittenden of Yale University are of interest.

The field of dietetics has not yet ceased to be a ground for controversy, and its history so far has been a disproving to-day of what was taken up yesterday. Prof. Chittenden happens to be the present authority, although the publication of the results of his experiments five years ago was the cause of a great upheaval.

"It is self-evident that the smallest amount of food that will serve to keep the body in a state of high efficiency is physiologically the most economical, and hence the best adapted to the needs of the organism," says Prof. Chittenden in his book "Physiological Economy of Nutrition."

To find this smallest amount of food required by three distinctly different types of individuals he undertook a series of experiments continuing over several months. The results of these experiments he gives in his book.

In the first group were five men, of different ages, professions and instructors in Yale University, representing the mental rather than physical worker. In the second group was a detail of volunteers from the Hospital Corps of the United States Army, representing the moderate worker, men of different nationalities, ages and temperaments. The third group was composed of eight young men, Yale students, trained athletes, some of them with exceptional athletic records.

The professor, in describing the experiments which he conducted with himself as a member of the first group says: "I weighed 45 kilos, was nearly 47 years of age, and accustomed to eating daily an amount of food approximately equal to the so-called dietary standards. Realizing that the habits of a lifetime should not be too suddenly changed, a gradual reduction was made in the amount of protein food taken each day, resulting in a month or so in the complete abolition of breakfast, except for a small out-of-door lunch was taken at 1:30 P. M., followed by a heavier dinner at 7:30. The total intake was diminished as well as the protein constituents."

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both body and mind are fully satisfied with the smaller amounts of food."

Here is a sample day's bill of fare for the soldiers:

BREAKFAST: Boiled Indian meal, 150 grams; milk, 135 grams; sugar, 30 grams; coffee, 1 cup; baked potato, 100 grams; butter, 10 grams.

DINNER: Clam chowder with onions, potatoes and tomatoes, 200 grams; bread, 75 grams; mashed potatoes, 100 grams; coffee, 1 cup.

SUPPER: Apple fritters, 200 grams; jam, 75 grams; tea, 1 cup; gingerbread, 30 grams.

In regard to his experiments with the

third group Prof. Chittenden says: "Men in training for athletic events deem it necessary to consume large amounts of protein food. Custom and long experience sanction a high protein diet composed largely of meat or of other food-stuffs rich in nitrogen for the development of that vigor and strength that go to make the accomplished athlete; but the daily use of such quantities of protein food as usually enter into the diet of the average athlete is very questionable."

"The subjects were the best athletes in all lines that Yale could produce. In regard to the experimentation Prof. Chittenden says: "It is quite evident from a study of the results obtained in the foregoing experiments that young, vigorous men of the

type under observation, trained in athletics, accustomed to the doing of vigorous muscular work, can satisfy all the true physiological needs of their bodies and maintain their physical strength and vigor as well as their capacity for mental work with an amount of protein food equal to one-half or one-third that ordinarily consumed by men of this stamp. It is clear that physiological economy in nutrition is as safe for men in athletics as for men not accustomed to vigorous exercise."

"The athlete, as well as the less active man physically, or the professional musician meet all his ordinary requirements with an intake of protein food far below the quantities generally consumed, and the value obtained in the foregoing experiments that young, vigorous men of the

amount of non-nitrogenous food.

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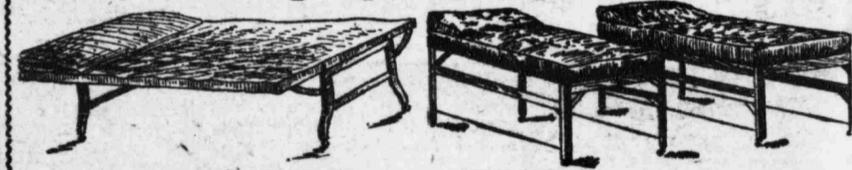
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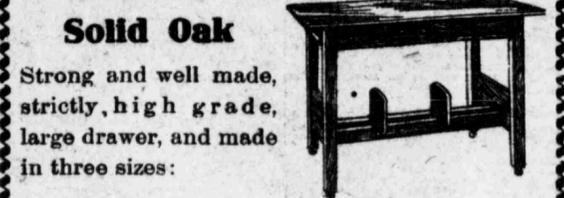
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Indians and Virginia to Meet at Football.

It has been learned that the Carlisle Indians will play football in Washington next fall against the team of the University of Virginia, which will not meet Georgetown. The loss of the annual game with Virginia has hit the Washingtonians hard because it arose because the Washingtonian faculty has called football off for the time being, and the Virginia folks didn't want to take chances on not having a big game at all. Unless football is not played at all at Georgetown, it is said it is hard to figure out where the Indians and Carlisle will meet on November 5. George Washington University will not play its football playing fields if they have games scheduled themselves.